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Atoms, elements,
and molecules, Part
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Questions Give
examples of
monosaccharaides,
disaccaraides, and

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polysaccharaides.

Compare and contrast the following

polysaccharaides:

Glycogen, starch, chiton, cellulose

What is the monomer that makes up each of the following classes of biomolecules?

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The Chemistry of
Life (Biochemistry)

Homework/Study

Guide. list the
three subatomic
particle that
compose atoms.

The atoms in a
compound are held
together by
chemical bonds.

~~The Chemistry of~~

Page 13/28

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~~Life (Biochemistry)~~
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Unit 3: Chemistry
of Life Study Guide

1) What are the
major components
of an atom? Proton,
Electron and
Neutron 2) Which
Subatomic particle
is most important
to biological
systems? The most

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important
subatomic particle
to biological

systems is 3) Be
able to indentify
various Chemicals
bonds.

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Chemistry of Life:
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Chapter 2 - The
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Chemistry Of

solution with a high concentration of H ions is (very acidic/very basic).

2. Chemical reactions that absorb more energy than they release are (endergonic/exergonic). 3.

~~Chemistry of Life
Study Guide.docx
Chapter 2 \u2013~~

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AP Biology Unit 1-
The Chemistry of
Life. These are

important
functional group to
know. Complete
the following chart
of functional
groups. Name of
Group Structural
Formula Molecular
Formula. Amino.
Methyl. Carboxyl.

Access PDF Chemistry Of

Hydroxyl.
Sulfhydryl (thiol)
Phosphate. AP

Biology Unit 1- The
Chemistry of Life.
Proteins

~~AP Biology Unit 1-~~
~~The Chemistry of~~
~~Life~~

Biochemistry is the
study of chemical
processes within
and relating to

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living organisms. It deals with the structures, functions and interactions of biological macromolecules such as proteins, nucleic acids, carbohydrates and lipids. These molecules form the structure of cells and perform many

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of the functions
associated with
life.

Key

~~Chemistry of Life—
Anatomy &
Physiology~~

Chapter 2 The
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Study Guide. Terms
in this set (85) The
Main source of
energy for living
things.

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Carbohydrates.

Help carry out
chemical reactions.

Proteins. Contain
Hydrogen, oxygen,
nitrogen,
phosphorous, and
carbon. Nucleic
Acids. Transport
substances in and
out of cells.

Proteins.

Composed of
amino acids.

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Unit 2 CHAPTER 6

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below, write true or

false. true 1.

Carbon atoms can

bond together in

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straight chains,
branched chains,
or rings. false 2.

Large molecules
containing carbon
atoms are called
micromolecules.
false 3.

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