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Chemical changes occur when bonds are broken and/or formed between molecules or atoms. This means that one substance with a certain set of properties (such as melting point, color, taste, etc) is turned into a different substance with different properties. Chemical changes are frequently harder to reverse than physical changes.

3.6: Changes in Matter-Physical and Chemical Changes ...

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the substance being dissolved. physical change. *change in matter that mat alter the size,shape,or phase but the atoms are not rearranged to form new substances. chemical change. *change in which atoms are rearranged to form a new substance.

Chemistry: Review of Matter Vocabulary Flashcards | Quizlet

Matter undergoes phase changes or phase transitions from one state of matter to another. Below is a complete list of the names of these phase changes. The most commonly known phase changes are those six between solids, liquids, and gasses. However, plasma also is a state of matter, so a complete list requires all eight total phase changes.

List of Phase Changes Between States of Matter

The atoms in molecules, crystals, metals and diatomic gases—indeed most of the physical environment around us—are held together by chemical bonds, which dictate the structure and the bulk properties of matter. Examples of Lewis dot -style representations of chemical bonds between carbon (C), hydrogen (H), and oxygen (O).

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